



## Nobel Prize for Chemistry 2021

### Why in news?

Chemistry Nobel 2021 was awarded to German scientist Benjamin List and Scotland-born scientist David W.C. MacMillan who independently developed a new way of catalysis – **asymmetric organocatalysis** in 2000.

### What are catalysts?

- When two or more compounds react to form new compounds, the process is often aided by other chemicals that do not change themselves, but help speed up the reaction. Such chemicals are called catalysts.
- Till around 2000, only two kinds of chemicals were known to act as effective catalysts - heavier metals and enzymes.
- **Heavier metals** - Heavier metals are expensive, difficult to mine, and toxic to humans and the environment.
- Also traces remain in the end product affecting the purity of compound like medicines.
- Metals required an environment free of water and oxygen, which was difficult to ensure on an industrial scale.
- **Enzymes**- Enzymes work best when water is used as a medium for the chemical reaction but that is not suitable for all kinds of chemical reactions.

### What is asymmetric organocatalysis?

- Substances having same chemical composition and molecular formula but different properties are known as **isomers**.
- Two molecules could be exactly the same, except that they are mirror images of each other which are often referred as left-handed or right-handed molecules.

# 2021 NOBEL PRIZE IN CHEMISTRY



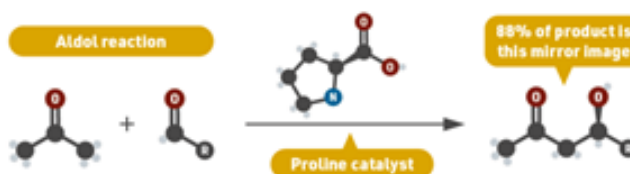
Awarded jointly to **Benjamin List** and **David W.C. MacMillan** for their development of a new type of catalysis, asymmetric organocatalysis, a precise new tool for molecular construction.



Mirror image molecules (enantiomers)

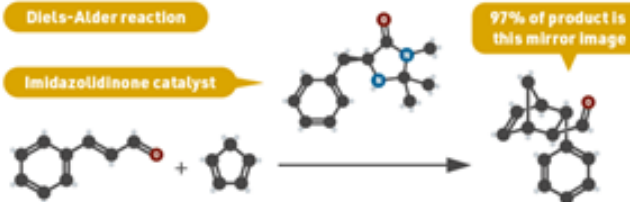
Molecules can exist in mirror image forms with different properties. To make medicines and other compounds we often want just one of these forms. Catalysts, substances that speed up reactions, can help. Nature's catalysts, enzymes, build specific mirror image molecules all the time, but have complicated structures. Metal catalysts can do the job but are sensitive to air and moisture, and sometimes harmful to the environment.

**Benjamin List** wondered if the single amino acids which build up enzymes could catalyse a reaction on their own. Knowing of previous research on the amino acid proline acting as a catalyst, he used it to catalyse an aldol reaction and found it was efficient – and also formed one mirror image of the product much more often than the other.



Diels-Alder reaction

Imidazolidinone catalyst



**David MacMillan** tried to develop alternatives to metal catalysts using organic molecules. He identified an imidazolidinone molecule which could catalyse a carbon-carbon bond forming reaction and produced mainly one mirror image of the product. He coined the term 'organocatalysis' for the concept of catalysing reactions using small organic molecules.



Cheaper to use



Environmentally friendly



More efficient

## WHY DOES THIS RESEARCH MATTER?

The winners, along with other researchers, have since designed many more organic molecule catalysts. They are cheap, environmentally friendly, and can make specific mirror images of molecules. They can be used one after another for different reaction steps, improving the efficiency of molecule-making.

Nobel Prize in Chemistry press release: <https://www.nobelprize.org/prizes/chemistry/2021/press-release/>



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- List and MacMillan discovered that by using a natural compound like an **amino acid as a catalyst**, only **one specific mirror image** of the end-product could be obtained. This was named as asymmetric catalysis.

## Why is the discovery so significant?

- New organocatalysts developed have helped drive a variety of chemical reactions in turn accelerating pharmaceutical drug research.
- The organic catalysts are cheap, environment friendly and are more efficient.
- The asymmetric organocatalysts have allowed researchers to efficiently produce new molecules with complete certainty of the 3-D orientation.

- Asymmetric organocatalysts allow several steps in molecule production to continue without interruption, minimising waste.
- Asymmetric organocatalysts were used to rectify the birth defects caused by thalidomide (used to treat nausea in pregnant women) by producing molecules of the desired mirror image form.
- This year's Prize underlines the message that great discoveries can come even from simple ideas which are often overlooked.

**Source: The Hindu, The Indian Express**



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